

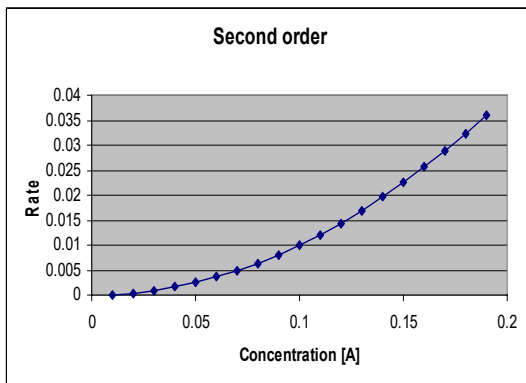
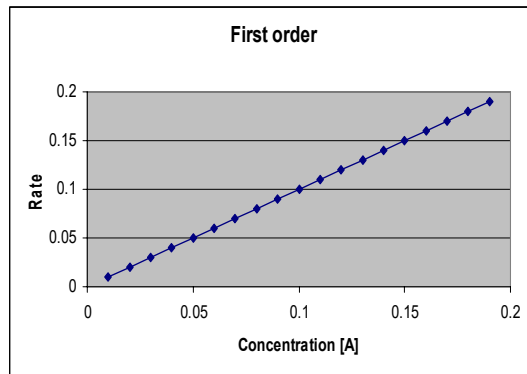
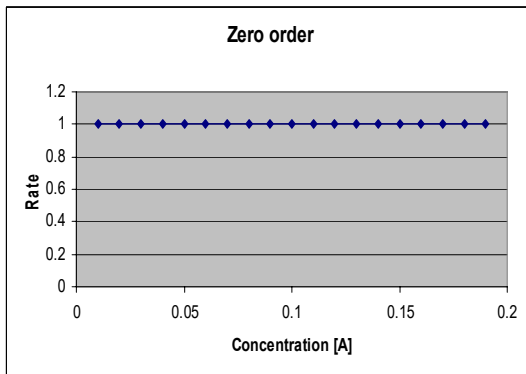
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Guided Inquiry Activity
Enzyme Kinetics, Part One
Answers

Model 1 Critical thinking questions

1. Decreases
2. It is being converted to B.
3. Decreases
4. As the concentration of A falls, the rate of conversion to B also decreases.
5. $1\text{M}/20\text{ minutes} = 0.05\text{ M/min}$
6. $0.7\text{ M}/5\text{ minutes} = 0.14\text{ M/min}$
7. $0.2\text{ M}/5\text{ minutes} = 0.04\text{ M/min}$
8. Yes, identified that rate as decreasing with time.
9. Use a tangent line or take a derivative.

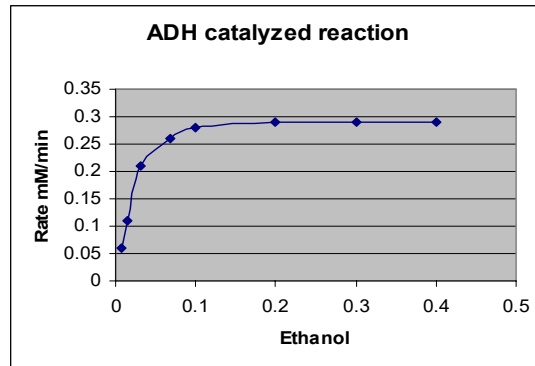
Graphs:



Questions related to above graphs:

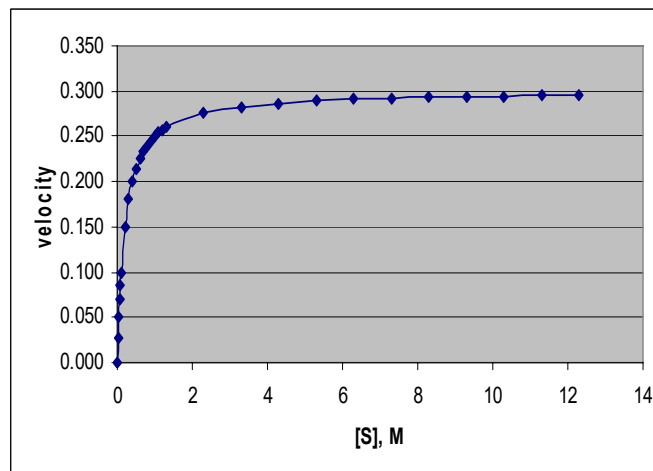
2. The rate is independent of concentration
4. The rate increases linearly with concentration
6. The rate increases exponentially with concentration.
7. As the rate constant doubles, all rates double.

Model 2

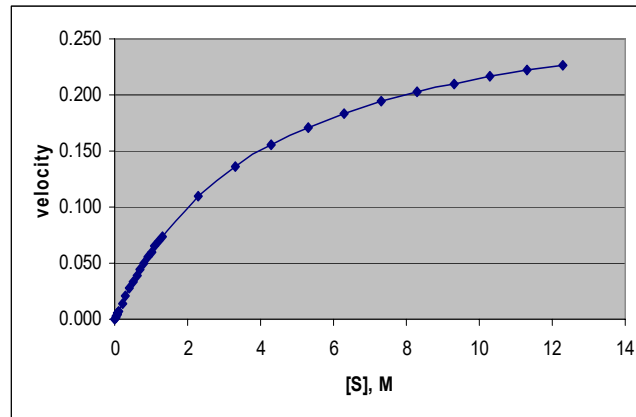


Critical Thinking questions

1. The shape is hyperbolic
2. The reaction follows first order kinetics up to about 0.031 M ethanol.
3. Zero order kinetics are observed between 0.2 and 0.4 M ethanol.
4. V_{\max} is 0.29 mM/min.
5. The enzyme becomes rate limiting; all active sites are filled.



1. Yes, the Michaelis Menton model produces the expected hyperbolic curve.
2. The asymptote is 0.3, V_{\max} .
3. The velocity slows, V_{\max} is unchanged, but is not reached over this substrate interval.



1. $V_{\max}/0.5 V_{\max} = 2$, $K_m = [S]$ at $\frac{1}{2} V_{\max}$, so it has units of concentration typically M or mM.
2. $0.5 V_{\max}$ is 0.15 when V_{\max} is 0.3.
3. The perpendiculars will give values of K_m as 0.2 or 4.
4. V_{\max} is approximately 2.3. $0.5 V_{\max}$ is then 1.15. Dropping a perpendicular to the x axis here gives a K_m of 0.8 – 0.9.
5. The enzyme that takes less substrate to reach half maximal velocity is more efficient. Hence, the enzyme that has a K_m of 0.2 is more efficient than an enzyme with a K_m of 4.0.